

**Bapatia Engineering College:: Bapatia(Autonomous)**

Department of Chemistry

Mid-Term Examination-II

L/4 B.Tech ECE,CSE ,IT and EIE

Engineering Chemistry

20EC103/20EC3102/20MT-102/20E1102

Time: 11:15 AM-12:45 P.M

Max. Marks: 35

30-03-2022

Question No.1 compulsory (1X7 = 7 M) ; Answer ONE Question from each unit (2X14 = 28 M)**1. Answer all questions****(1X7 = 7 Marks)**

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|---|-----|-----|----|
| a. Name the highest ranked coal | CO3 | BL2 | 1M |
| b. What are the units of calorific value | CO3 | BL1 | 1M |
| c. Illustrate the examples of the anti-knocking agent | CO3 | BL1 | 1M |
| d. What are the advantages of CNG | CO3 | BL2 | 1M |
| e. Describe biodegradable polymers? Give example | CO4 | BL1 | 1M |
| f. Define addition reaction | CO4 | BL1 | 1M |
| g. Define thermoplastics | CO4 | BL3 | 1M |

UNIT – III

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|--|-----|-----|----|
| 2. a. Compose the construction and working of Bomb calorimeter with neat label diagram. | CO3 | BL3 | 7M |
| 2. b. Write short notes on LPG | CO3 | BL2 | 7M |
| (OR) | | | |
| 3. a. Explain briefly about octane number. | CO3 | BL2 | 7M |
| 3. b. Define flue gas. How the analysis of flue gas is done by Orsat's apparatus. Write its significance | CO3 | BL1 | 7M |

UNIT – IV

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| 4. Generate the method of synthesis, properties and applications of "Aspirin and Paracetamol" | CO4 | BL3 | 14M |
| (OR) | | | |
| 5. a. Synthesis, properties and applications of Polyhydroxybuterate-co- β -hydroxyvalerate (PHBV) | CO4 | BL3 | 7M |
| 5. b. What is Markownikoff's of rule? Explain the mechanism | CO4 | BL1 | 7M |