Hall Ticket Number:



I/IV B.Tech (Regular) DEGREE EXAMINATION

| | | I/IV B. Tech (Regular) DEGREE EXAMINA | non | | | | |
|-------------------------------------|-------------------------------------------------------------------------|------------------------------------------------------------------------|-----------------------------------|------------------------------|--------------------|------------|--|
| APRIL, 2019 | | | | CE/ME/EEE Branches | | | |
| Second Semester | | | | Engineering Chemistry | | | |
| Time: Three Hours | | | | Maximum : 50 Marks | | | |
| Answer Question No.1 compulsorily. | | | | | (1X10 = 10 Marks) | | |
| Answer ONE question from each unit. | | | | (4X10=40 Marks) | | | |
| 1. Answer all questions | | | | Ŧ | , | =10 Marks) | |
| No. | Questions 1 (a to j) | | | | Level | COs | |
| a | - | Define hardness of water? | | | nember | CO 1 | |
| b | Define priming. | | | Remember | | CO 1 | |
| С | | What is meant by phosphate conditioning | | | lerstand | CO 1 | |
| d | What is galvanic corrosion? | | | Understand | | CO 2 | |
| e | | State 'Pilling-Bedworth rule''? | | | nember | CO 2 | |
| f | | Define Octane number. | | | nember | CO 3 | |
| g | | What is lower calorific value of fuel | | | lerstand | CO 3 | |
| h | | Write the composition of Petrol | | | nember | CO 3 | |
| i | - | Define addition reactions? | | Remember | | CO 4 | |
| j | Write | Write any two applications of Poly HydroxyButerate (PHB). | | | Apply | CO 4 | |
| | | UNIT – I | | | 1 | 1 | |
| No. | Ques | tions (2 to 9) | Leve | | COs | Marks | |
| | (a) | Define Alkalinity of water. Discuss the estimation of alkalinity | Remember, | | CO 1 | 6M | |
| | (a) | of water. | Unders | erstand | | | |
| 2 | (b) | 100 mL of water sample on titration with N/50 H_2SO_4 required | Understand, apply | | CO 1 | 4M | |
| | | 8.0 mL of the acid to phenolphthalein end point and 9 mL of | | | | | |
| | | the same acid to methyl orange end point. Determine the type | | | | | |
| | | and amount of alkalinity present in water sample. | | | | | |
| | | OR | | | 1 | | |
| | (a) | Explain the following i).Boiler corrosion ii). Caustic | Understand | | CO 1 | 6M | |
| 3 | (u) | embrittlement | | | | 0111 | |
| C | (b) | Discuss the method of treatment of brackish water by Electro | Underst | | | 4M | |
| | (-) | dialysis | apply | | | | |
| | | UNIT – II | | | 1 | | |
| | (a) | (a) Deduce Nernst equation for single electrode potential. Write | | lyze, CO | | 4M | |
| | () | its applications (any two) | Apply | | | | |
| 4 | | What is electrochemical corrosion? Describe the mechanism | | | CO 3 | | |
| | (b) of electrochemical corrosion by evolution of hydrogen type Under | | | | | 6M | |
| | and absorption of oxygen type. | | | | | | |
| | | OR | XX 1 | . 1 | | - 01 | |
| 5 | | (a) Discuss various factors affecting corrosion rate of a metal. Under | | | CO 2 | 6M | |
| | (b) Write short note on Electro plating of Gold. Under | | tand | CO 2 | 4M | | |
| | 1 | UNIT – III | | | 1 | | |
| 6 | | Define calorific value of a fuel. Discuss the construction and | Remember, Create Understand | | CO 3 | 7M | |
| | (a) | working of Bomb calorimeter to determine calorific value of | | | | | |
| | | solid fuel. | | | | | |
| | (b) | Write short note on knocking and anti-knocking agents. | Unders | tand | CO 3 | 3M | |
| | | OR | | | 1 | | |
| | (a) (b) | What is meant by Flue gas? Explain the method of analysis of | Remember, Understand | | CO 3 | 7M 3M | |
| 7 | | flue gas by Orsat apparatus | | | | | |
| | | What is LPG? What are the advantages of LPG over other | Remen | mber CO 3 | | | |
| | . / | gaseous fuels | | | | | |
| | | | TT 1 | | | | |
| 8 | (a) | | | | CO 4 | 7M | |
| | (b) Describe a method of synthesis of "Aspirin" Under | | tand | CO 4 | 3M | | |
| | 1 | OR | | | 1 | | |
| 9 | (a) Distinguish between Thermoplastic and Thermosetting Analy | | | ze | CO 4 | 6M | |
| | | polymers | | | | | |
| | (b) Explain the preparation and applications of Bakelite | | | tand | CO 4 | 4M | |